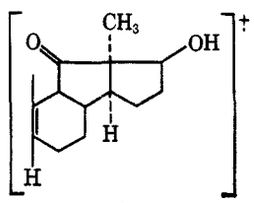
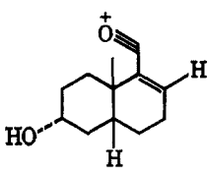
Page 3628. Structure 8 in Scheme I should have the same stereochemistry for the methyl groups as shown for 6 and 7.

Page 3632. Structures q and t in Scheme III should be as follows.



q, *m/e* 220



t, *m/e* 193

A. H. Weinstein: Sulfonation–Debutylation of 2,6-Di-*t*-butyl-*p*-cresol.

Page 3669. Column 2, Scheme I. Formula II should be the 2-chlorosulfonylphenol listed as compound II in the text and not a 2-chlorosulfonyl aromatic sulfonic acid.

Stanley R. Sandler: Reactions of *gem*-Dihalocyclopropanes with Electrophilic Reagents. Formation of Allyl Derivatives and/or Dienes.

Page 3878. Column 2. In Scheme II, the ion B is missing the bromine atom at C-3.

Page 3880. In Table IV, for XI the structural assignment opposite δ = 1.85 should be as follows.

$$CH_3—\overset{|}{C}=C<$$

Ronald A. LeMahieu: The Reaction of Di-*n*-butylcadmium with Derivatives of Ketal Acids.

Page 4150. Column 1. The legend for formula I should read R = Cl or $O\overset{O}{\overset{\|}{C}}OC_2H_5$. The legend for formula Vb should read R = $CO_2C_2H_5$. The legend for formula IXb should read R = $CO_2C_2H_5$.

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C. A. Bunton, S. J. Farber, and Eleanor J. Fendler: The Hydrolysis of p-Nitrophenyl Diphenyl Phosphate.

Page 31. Table V, line 9. 10^4k_ψ = 172 sec^{-1}. The corresponding value of k_2 is correct.

G. Grethe, H. L. Lee, M. Uskoković, and A. Brossi: Syntheses in the Isoquinoline Series. Synthesis and Chemical Transformation of 2,3-Dihydro-4(1H)-isoquinolones.

Page 496. The title of Figure 1 should read as follows.

Figure 1.—Infrared absorption in the region of 1400–2000 cm^{-1} for the salts of the keto esters **30** (A), **32** (B), and **34** (C) taken in KBr pellets.

G. Grethe, V. Toome, H. L. Lee, M. Uskoković, and A. Brossi: Syntheses in the Isoquinoline Series. Selective Demethylation of 6,7- and 7,8-Dimethoxy-2,3-dihydro-4(1H)-isoquinolones.

Page 505. The title of Figure 1 should read as follows.

Figure 1.—Ultraviolet spectra of 2-benzyl-2,3-dihydro-6-hydroxy-7-methoxy-4(1H)-isoquinolone hydrobromide (**9b**) in ethanol (—), in ethanol saturated with sodium acetate (---), and in 0.002 *M* ethanolic sodium ethoxide (···).

Page 505. The title of Figure 2 should read as follows.

Figure 2.—Ultraviolet spectra of 2-benzyl-2,3-dihydro-7-hydroxy-6-methoxy-4(1H)-isoquinolone hydrochloride (**10**) in ethanol (—), in ethanol saturated with sodium acetate (---), and in 0.002 *M* ethanolic sodium ethoxide (···).

Page 506. The title of Figure 3 should read as follows.

Figure 3.—Ultraviolet spectra of 2,3-dihydro-7-hydroxy-6-methoxy-1(4H)-naphthalenone (**15**) in ethanol (---) and ethanol saturated with sodium acetate (—).

Page 506. The title of Figure 4 should read as follows.

Figure 4.—Ultraviolet spectra of 2,3-dihydro-6-hydroxy-7-methoxy-1(4H)-naphthalenone (**16**) in ethanol (---) and ethanol saturated with sodium acetate (—).

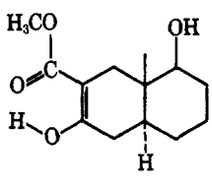
James W. Wilt, Charles F. Parsons, Charles A. Schneider, David G. Schultenover, S. J., and William J. Wagner: The Preparation and Study of Some 1-Norbornenyl and Norbornenyl-1-carbinyl Derivatives.

Page 694. The name of the second author should be Charles F. Parsons.

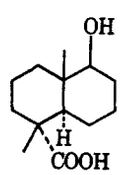
Page 699. Column 1. The ΔS^* values in Table I are in error. The value for tosylate **42** is -17.0 ± 0.1 eu. The value for tosylate **38** is -12.0 ± 0.1 eu. The discussion presented in the paper is affected by this correction to the extent that the inductive retardation by the double bond in **38** is reflected not only in enthalpic differences but also in entropic ones. We view this as an indication of less charge development in the activated complex from **38** compared with that from **42**. This is in line with our view that the transition state is reached later from **38** than from **42**, as we stated in the paper.

Thomas A. Spencer, Thomas D. Weaver, Rodolfo M. Villarica, Richard J. Friary, Jeanette Posler, and Martin A. Schwartz: Syntheses of Methyl Deisopropyldehydroabietate. Diterpenoid Synthesis by the AB → ABC Approach.

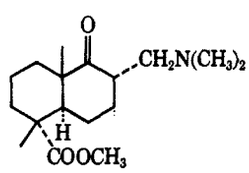
Page 713. Column 2. Structural formula **11** should be as shown.



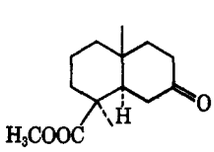
Page 714. Column 1. Structural formula **19** should be as shown.



Page 714. Column 2. Structural formula **25** should be as shown.



Page 714. Column 2. Structural formula **39** should be as shown.



T. Uematsu and R. J. Suhadolnik: 7-Deazaadenine Ribonucleosides. The Use of Periodate Oxidation in Degradation Studies.

Page 726. The first paragraph of this paper, which should have been published as a Note, was inadvertently published as an abstract.

M. S. Chang and J. U. Lowe, Jr.: Di(cyclopropanecarbonyl)-furoxan.

Page 866. Paragraph 3, line 3. After colorless, insert, "solid and a yellow solid reminiscent of the formation of a colorless." In formula II, change exocylic N—O bond to dative N→O. In formula IV, insert double bond in positions 4–5 of isoxazole ring.

John Jacobus, Morton Raban, and Kurt Mislow: The Preparation of (+)-N-Methyl-1-(1-naphthyl)ethylamine and the Determination of Its Optical Purity by Nuclear Magnetic Resonance.

Page 1142. Footnote 8. "Toluenesulfonates" should read "methanesulfonates."

Page 1144. The rotation of (+)-N-methyl-1-(1-naphthyl)-ethylamine in ethanol is strongly dependent on the water content of the ethanol. In rigorously dried ethanol (stored over Linde 4A Molecular Sieves), the rotation is $[\alpha]^{24}D$ +89° (*c* 3.51). The value originally reported, $[\alpha]^{25}D$ +74° (*c* 3.885), refers to *ca.* 90% ethanol.

B. Franzus, W. C. Baird, Jr., and J. H. Surridge: Synthesis of *exo,exo*-5,6-Dideuterio-*syn*-7-acetoxynorbornene and *exo,exo*-5,6-Dideuterio-2-norbornene.